This article was downloaded by:

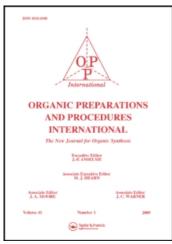
On: 27 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t902189982

4-METHOXYBENZENETHIOL

- J. Szmuszkovicz^a
- ^a Research Laboratories, The Upjohn Company, Kalamazoo, Michigan

To cite this Article Szmuszkovicz, J.(1969) '4-METHOXYBENZENETHIOL', Organic Preparations and Procedures International, 1: 1, 43-45

To link to this Article: DOI: 10.1080/00304946909458347 URL: http://dx.doi.org/10.1080/00304946909458347

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

4-METHOXYBENZENETHIOL

J. Szmuszkovicz Research Laboratories The Upjohn Company, Kalamazoo, Michigan 49001

$$CH_3O$$
 $-MgBr + S \longrightarrow CH_3O$
 $-SH$

We would like to bring to your attention a convenient and efficient procedure for the preparation of 4-methoxybenzenethiol by the treatment of p-methoxyphenylmagnesium bromide with sulfur¹, which has been overlooked in the literature in preference to other methods. Thus, 4-methoxybenzenethiol has been prepared by reduction of p-methoxyphenylsulfinic acid with zinc²; by reduction of p-methoxyphenylsulfonyl chloride with zinc³, tin⁴, red phosphorus and iodine⁵, or by electrolytic reduction⁶; and by diazotization of p-anisidine followed by treatment with potassium ethyl xanthate and alkaline hydrolysis.⁷

Experimental

p-Methoxyphenylmagnesium bromide A 5 1., three neck, round bottom flask, equipped with a condenser, thermometer, addition funnel and air-stirrer, was flame dried under nitrogen. Magnesium (38.8; 1.6 mole) was placed in the flask and covered with ether. One crystal of iodine was added, then about 20 ml of a solution containing 300 g. (1.6 mole) of p-bromoanisole in 1600 ml. of ether. Reaction started in a few minutes and the mixture was then stirred and refluxed while the rest of the above solution was

J. SZMUSZKOVICZ

added during about 1.5 hr. After the addition was completed, the mixture was refluxed for 1 hr.

The above mixture was cooled to 30°. Solid sulfur 4-Methoxybenzenethiol (46.4 g; 1.45 mole) was added portionwise over 30 min. with only occasional cooling so that the temperature was 30-35. The mixture was then stirred for 1 hr. at room temperature, cooled to 0 with a methanol-ice bath and decomposed by slow addition of 1600 ml. of 2.5 N hydrochloric acid keeping the temperature below 5°. The organic layer was separated⁸ and extracted with 2 N sodium hydroxide (5 x 200 ml.). The basic extract was cooled in ice and acidified with 650 ml. of 10% hydrochloric acid (check pH). The product was extracted with ether (4 x 200 ml.). The ether extract was washed with 200 ml. of saturated sodium chloride solution, dried (MgSO,) and evaporated to give 134 g. of residue. Distillation through a 15 cm. Vigreux at 13 mm gave 110.6 g. (49% yield) of an oil boiling at 100-103 (no forerun, some pot residue present). EtOH $_{\lambda max}$ 239 mu (10,000); 285.5 (1,300), sh 291 (1,250); $_{\text{vmax}}^{\text{Nujol}}$ SH: 2560; C=C: 1590, 1570, 1490; C-O: 1285, 1240, 1180, 1175, 1030; aromatic: 820, C-S: 635, 625. Nmr (CDC1₃) solution, 60-Mc, tetramethylsilane): SH singlet at 201 cps, area 1; OCH₂ singlet at 244 cps, area 3; aromatic = typical para substituted pattern centered at 408 and 437 cps.

References

- This procedure was reported briefly without experimental details by M. F. Taboury, Bull. Soc. Chim. France, [3] 33, 836 (1905).
- 2. L. Gatterman, Ber., 32, 1136 (1899).
- Y. Schaafsma, A. F. Bickel and E. C. Kooyman, Rec. Trav. Chim., 76, 180 (1957); L. Almasi, A. Hantz, and L. Paskucz, Acad. Rep. Populare Romine, Filiala Cluj., Studii Cercetari Chem., 12, 165 (1961); M. Protiva, M. Rajsner, E. Adlerova, V. Seidlova and Z. J. Vejdelek, Collection Czech. Chem. Commun., 29, 2161 (1964).

- 4. W. L. Nobles and B. B. Thompson, J. Pharm. Sci., <u>54</u>, 709 (1965).
- 5. A. W. Wagner, Ber., 99, 375 (1966).
- 6. F. Fichter and W. Tamm, Ber. 43, 3032 (1910).
- C. M. Suter and H. L. Hansen, J. Am. Chem. Soc., <u>54</u>, 4100 (1932);
 V. N. Vasileva and E. N. Guryanova, J. Gen. Chem. USSR (Engl. Transl.)
 <u>26</u>, 777 (1956); E. E. Campaigne, J. Tsurugi and W. W. Mayer, J.
 Org. Chem., <u>26</u>, 2486 (1961).
- 8. Dissemination of bad odor can be avoided by working in a good hood without spilling and letting all the used equipment soak in 5% sodium hydroxide solution overnight.

(Received August 9, 1968)